(357 mg, 0.67 mmol) in toluene (20 mL) afforded 8b (257 mg, 57%):  $R_1$ , 0.32 with ether-petroleum ether (1:1); mp 63-64 °C (methanol-ethyl acetate);  $[\alpha]_{D}$  = +8.3° (c 1.4, CHCl<sub>3</sub>); <sup>13</sup>C NMR 128.29, 128.24, 128.09, 127.92, 127.85, 127.73, 127.44, 127.2 (C arom), 109.94 (C1'), 98.11 and 96.75 (OCH<sub>3</sub>) 83.67 (C3'), 81.03 (C5'), 78.24 (C2'), 75.66, 74.84, 74.58, 73.08, 72.98 ( $-CH<sub>2</sub>Ar$ ), 73.29  $(C4')$ , 68.76 (C6'). Anal. Calcd for  $C_{43}H_{44}O_8$ : C, 75.06; H, 6.45. Found: C, 74.49; H, 6.48. (CDC13) 143.83, 138.84, 138.72, 138.53, 138.43, 130.81, 128.73,

1,1-Anhydro-1-C-[2-(hydroxymethyl)-4,6-dimethoxyphenyl]- $\beta$ -D-glucopyranose (9b). Hydrogenolysis of 8b (165) mg, 0.24 mmol) in methanol (1 mL) and ethyl acetate (1 mL) in the presence of 10% palladium on charcoal (20 mg) afforded 9b in quantitative yield (83 mg), which was purified by preparative TLC (68 mg, 87%): *R,* 0.40 with **dichloromethane-methanol**  (80:20); mp 97-98 °C (ethyl acetate);  $[\alpha]_D = +48.1$ ° (c 0.43, CHCl<sub>3</sub>); MS  $m/z$  329 (M + 1). Anal. Calcd for C<sub>15</sub>H<sub>20</sub>O<sub>8</sub>: C, 54.92; H, 6.15. Found: C, 54.89; H, 6.27.

**l,l-Anhydro-l-C-[2-(hydroxymethyl)-4,6-dimethoxy**phenyl]- $\beta$ -D-glucopyranose Tetraacetate (10b).<sup>2,3</sup> Acetylation of 9b (44 mg, 0.128 mmol) according to the procedure described for the preparation of 10a [acetic anhydride (128  $\mu$ L), pyridine (1 mL), and DMAP (1 crystal)] afforded **lob,** which was purified by preparative TLC (47 mg, 74%):  $R_f$  0.41 with petroleum eth-<br>er-ethyl acetate (1:1); mp 159–160 °C (petroleum ether) (lit.<sup>3</sup> mp 70-72 °C);  $[\alpha]_D = +7^\circ$  *(c 1, CHCl<sub>3</sub>)* [lit.<sup>3</sup>  $[\alpha]_D = +5^\circ$  *(c 1, CHCl<sub>3</sub>)*]; MS *m*/*z* 496 (M + 1); <sup>1</sup>H NMR (CDCl<sub>3</sub>) *δ* 6.30–6.28 (2s, 2, H arom), 5.89 (d, 1,  $J_{2',3'}$  = 9.98 Hz, H2'), 5.55 (t, 1,  $J_{3',2'} = J_{3',4'}$  = arom), 5.89 (d, 1,  $J_{2',3'} = 9.98$  Hz, H2'), 5.55 (t, 1,  $J_{3',2'} = J_{3',4'} = 9.84$  Hz, H3'), 5.30 (t, 1,  $J_{4',3'} = J_{4',5'} = 9.9$  Hz, H4'), 5.16 (d, 1,  $J = 12.7 \text{ Hz}$ , ArCH<sub>2</sub>O-), 5.03 (d, 1,  $J = 12.7 \text{ Hz}$ , ArCH<sub>2</sub>O-),<br> $J = 12.7 \text{ Hz}$ , ArCH<sub>2</sub>O-), 5.03 (d, 1,  $J = 12.7 \text{ Hz}$ , ArCH<sub>2</sub>O-), 4.28-4.07 (2 m, 3, H5', H6', H6"), 3.82 **(8,** 3, -OCH,), 3.78 **(8,** 3, for C23H28012: C, 55.69; H, 5.69. Found: C, 55.70; H, **5.74.**  CH<sub>3</sub>CO-), 1.99 (s, 3, CH<sub>3</sub>CO-), 1.73 (s, 3, CH<sub>3</sub>CO-). Anal. Calcd

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Registry **No.** 1,13096-62-3; 2a, 132814-55-2; 2b, 135877-95-1; 3a, 135877-99-5; 3b, 135878-00-1; 4a, 5333-62-0; 4b, 135877-96-2; 5a, 132814-51-8; 5b, 135877-97-3; 6,62641-00-3; 8a, 132814-52-9; 8b, 135877-98-4; 9a, 13281453-0; 9b, 76843-395; **loa,** 132814-54-1; 10b, 76843-40-8; 2-BrC<sub>6</sub>H<sub>4</sub>CH<sub>2</sub>OH, 18982-54-2; 2-Br-3,5- $(MeO)<sub>2</sub>C<sub>6</sub>H<sub>2</sub>CH<sub>2</sub>OH$ , 74726-76-4; papulacandin A, 61036-46-2; papulacandin B, 61032-80-2; papulacandin C, 61036-48-4; papulacandin D, 61036-49-5.

Supplementary Material Available: Crystallographic data for compound 10a (6 pages). Ordering information is given on any current masthead page.

# **Synthetic Applications of Protected 2-Aryl-4-piperidones. 7.' Synthesis of 1 -Et hylindolo[** *2,3-a* **]quinolizidin-2-one~**

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The synthesis of **l-ethylindolo[2,3-a]quinolizidin-2-one** (1) by the intramolecular cyclization of protected N-(2-hydroxyethyl)-2-[ **l-(phenylsulfonyl)-3-indolyl]-4-piperidone** 15 by the action of KtBuO **and** further acid treatment is reported. The methodology has been first carried out for ite deethyl analogue 2 from (hydroxyethyllpiperidine **14** and has shown to be a good general method to reach **indolo[2,3-a]quinolizidin-2-one** systems. Compound **1** has also been obtained by an unusual rearrangement in acidic medium of 7-ethylhexahydro $pyrido[1',2':1,2] pyrazino[4,3-a]indole.$ 

### **Introduction**

In continuing our studies of the synthesis of the indo-**10[2,3-a]quinolizidin-2-0ne** system2 **as** a new synthetic application of easily accessible protected 2-aryl-4 piperidones, $3$  we report now the synthesis of 1-ethyl**indolo[2,3-a]quinolizidin-2-one** (l), which can be considered a key intermediate in the preparation of pentacyclic indole alkaloids of the vincamine type,<sup>4</sup> some of which, such as  $(-)$ -eburnamonine and  $(+)$ -vincamine,<sup>5</sup> are used in medicine for their vasodilator properties (Scheme I).

In previous work, we already reported the synthesis of the **indolo[2,3-a]quinolizidin-2-one** basic framework **(2),2b**  as well as its 3-ethyl derivative,<sup>2a</sup> by the intramolecular cyclization of protected **N-(2-hydroxyethyl)-2-[l-(phenylsulfonyl)-2-indolyl]-4-piperidones** (3 and **4)** by the initial action of potassium  $tert$ -butoxide.<sup>6</sup> However, the cycli-



zation occurred as well, to some extent, upon the indole nitrogen atom, yielding the corresponding hexahydro-4H-

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<sup>&#</sup>x27;Faculty of Chemistry.

<sup>&</sup>lt;sup>1</sup>This work is dedicated to Professor Felix Serratosa for his long **and** impressive years of research.

<sup>(1)</sup> For part VI, see: Rubiralta, M.; Marco, P.; Bolós, J.; Trapé, J.<br>
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Table I. <sup>1</sup>H NMR Spectra Data<sup>a,b</sup> of 3-(2-Piperidyl)indoles

proton	14	15	21	22	23
$2-Ha$	$3.80$ dd $(12, 3)$	$3.44$ d $(11)$	4.15 dd (12, 2)	$3.86$ d $(12)$	4.43 dd (12, 0.8)
$3-He$	$1.76 - 1.89$ m		$1.75 - 1.85$ m		masked
$3-Ha$	2.21 t(12)	masked	1.81 t $(12)$	$1.60 - 1.80$ m	
5-He	$1.76 - 1.89$ m	1.84 br d $(12)$	$2.03$ br d $(12)$	$1.95$ dt $(11, 5)$	1.56 ddd (12, 4, 1.7)
$5-Ha$	$2.09$ td $(12, 3)$	$2.05$ br t $(12)$	$1.75 - 1.85$ m	$1.60 - 1.80$ m	$1.87 \text{ td } (13, 5)$
6-He	$3.20 - 3.30$ m	$3.10 - 3.20$ m	$3.20$ dt $(12, 4)$	$2.97 - 3.09$ m	$3.12 \;\mathrm{br}$ d $(12)$
6-Ha	$2.45$ td $(12, 3)$	2.37 <sub>m</sub>	$3.02$ ddd $(13, 12, 5)$	$2.92$ td $(11, 5)$	$2.98$ td $(12, 3.5)$
OCH <sub>2</sub> CH <sub>2</sub> O	$3.90 - 4.00$ m	$3.90 - 4.00$ m	$3.90 - 4.00$ m	$3.90 - 4.10$ m	$3.90 - 4.00$ m
CH <sub>2</sub> CH <sub>3</sub>		$0.70 - 0.90$ m		$0.81 - 1.00$ m	$1.28 - 1.41$ m
		$1.20 - 1.40$ m		$1.20 - 1.40$ m	
$CH_2CH_3$		0.21 t(7)		0.35 t(7)	0.17 t(7)
$In-2H$	7.54 s	7.52s	$7.55$ s	$7.55$ s	7.55 s
In-4H	7.70 d(7)	7.75 d(7)	7.63 d(7)	7.80 d(7)	$7.20 - 7.55$ m
$In-5H$	$7.20 - 7.50$ m	$7.20 - 7.50$ m	$7.30 - 7.50$ m	$7.20 - 7.45$ m	$7.20 - 7.50$ m
$In-6H$	$7.20 - 7.50$ m	$7.20 - 7.40$ m	$7.30 - 7.50$ m	$7.20 - 7.45$ m	$7.20 - 7.50$ m
$In-7H$	7.97 d(7)	8.00 d(7)	8.00 d(7)	8.00 d(7)	8.00 d(7)
$NCH_2CH_2$	$2.70$ and $3.23$ m	2.57 and 3.11 m			
OCH <sub>2</sub>	$3.56$ td $(11, 4)$	$3.42 - 3.53$ m			

<sup>a</sup> Recorded at 200 MHz in CDCl<sub>3</sub>. J values in brackets are reported in hertz and chemical shifts are given in  $\delta$  units (downfield from Measi). bPhenyl signals: 7.26 t (7), 7.45 t (7), 7.86 d (7) for 14; 7.20-7.40 m, 7.85 d (7) for **15;** 7.26 t (7), 7.42 t (7), 7.88 d (7) for 21; 7.21-7.49 m, 7.84 d (7) for 22; 7.38 t (7), 7.48 t (7), 7.87 d (7) for 23.

Table **11.** *\*c* **NMR** Spectral Data" of 3-(2-Piperidyl)indoles

carbon	14	15	21	22	23
C-2	59.6	61.1	53.3	56.5	56.1
C-3	40.9	47.2	40.3	51.6	47.4
C-4	104.4		105.0	110.9	110.0
C-5	32.1	30.6	31.7	36.2	31.5
C-6	53.9	49.8	43.6	44.2	43.9
OCH <sub>2</sub>	59.0	63.8	64.7	64.9	64.0
		64.3	65.1	65.5	64.2
CH <sub>2</sub> CH <sub>3</sub>		18.8		19.1	17.6
CH <sub>2</sub> CH <sub>3</sub>		14.9		14.8	14.5
$In-C2$	138.5	137.0	137.4	144.0	143.0
In-C3	114.9	112.5	112.0	109.4	110.5
In-C3a	128.8	129.1	129.4	130.0	128.9
In-C4	124.7	123.9	124.4	110.9	124.0
In-C5	122.6	120.8	122.1	121.5	120.7
In-C6	126.7	124.6	125.6	125.3	124.6
In-C7	116.1	115.3	114.8	115.6	115.4
In-C7a	132.9	137.0	136.9	138.5	139.0
$(C_6H_5)$ o	126.2	126.6	127.1	127.9	126.4
$(C_{6}H_{6})m$	129.6	129.2	129.8	129.8	129.2
$(C_6H_5)p$	134.6	134.0	134.4	134.5	133.8
$(C_6H_6)$ i	137.2	139.0	136.2	140.0	138.0
$NCH_2CH_2$	56.7	54.5			
NCH.CH.	59.0	58.1			

<sup>a</sup> Recorded at 50.3 MHz in CDCl<sub>3</sub>. Assignments were aided by DEPT sequence experiments. Chemical shifts are given in *b* units (downfield from Me,Si).

pyrido[ **1',2':1,2]pyrazino[4,3-a]indoles 9** and **10,** depending on the solvent and the substrate. **Thus,** when the synthesis of **1-ethylindoloquinolizidine 8** was envisaged by the same strategy, the K<sup>t</sup>BuO reaction only provided the pyridopyrazinoindole regioisomer **11** in 76% yield (see Scheme **11).** 

Such inconvenience has been successfully circumvented by taking advantage of the known capability of 3,3-disubstituted indolenines to rearrange into 2,3-disubstituted indoles? Thus, the intramolecular cyclization of con-





Scheme **III<sup>a</sup>** 



'Reagenta and conditions: (i) benzene, Dean-Stark; (ii) *p-*TsOH, benzene, Dean-Stark; (iii)  $BrCH_2CH_2OH$ , EtOH,  $Na_2CO_3$ ; (iv) Et<sub>2</sub>O, 0 °C, 'BuOK; (v) LiAlH<sub>4</sub>, THF; (vi) 10% HCl, MeOH,  $\Delta$ .

veniently protected 2-[ **l-(phenylsulfonyl)-3-indolyl]-4**  piperidones **14** and **15** by the action **of** K'BuO has been

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**Table 111. 'H NMR Spectral Data' of Spiroindolines** 24-29



'Recorded at 200 MHz in CDC13. J values in brackets are reported in hertz and chemical shifts are given in **6** units (downfield from Me<sub>4</sub>Si).

**Table IV. NMR Spectral Data' of Spiroindolines** 24-29

carbon	24	25	26	27	28	29
C-1	39.0	38.8	46.2	47.9	48.1	$41.8^{b}$
$C-2$	49.8	49.9	50.4	49.4	49.3	50.7 <sup>b</sup>
$C-4$	52.8	52.8	52.5	52.0	53.2	$51.6^{b}$
C-5	34.4	34.4	31.1	34.3	34.6	40.7 <sup>b</sup>
C-6	109.0	108.0	110.5	110.9	111.5	211.2
C-7	35.8	34.7	43.0	42.3	44.6	53.5
C-8	71.5	71.6	73.8	72.7	76.4	74.9
C-9	59.0	57.3	59.2	56.6	61.1	56.8
$C-10a$	151.0	151.5	149.5	150.7	150.9	151.2
C-11	109.4	109.4	109.4	109.2	109.3	109.6
$C-12$	125.1	123.0	122.7	122.9	125.2	123.9
C-13	118.9	118.7	118.9	118.5	118.7	118.9
C-14	127.6	127.8	127.6	127.7	127.6	128.5
C-14a	128.5	133.9	135.5	135.8	135.0	134.2
C-14b	54.0	53.5	53.1	54.2	54.1	54.7
OCH,	64.1	64.2	64.2	64.3	64.2	
	64.2	64.3	64.4	64.8	64.8	
$CH_2CH_3$			21.9	17.2	17.3	17.7
$CH_2CH_2$			15.2	14.4	14.2	11.8

 $4^{13}$ C NMR spectra were recorded at 50.3 MHz in CDCl<sub>3</sub>. The assignments were aided by **DEPT** sequence experiments. Chemical shifts are given in  $\delta$  units downfield from Me<sub>4</sub>Si. <sup>b</sup>These signals can be interchanged.

applied to obtain the corresponding 3-spiroindolenines **12**  and **13,** whose rearrangement would constitute a new way to prepare **indolo[2,3-a]quinoliidin-2-one** systems, and in particular, rearrangement of 3-spiroindolenine **13** would provide **l-ethylindolo[2,3-a]quinolizidin-2-one (l),** our target compound.

#### **Results and Discussion**

The synthesis of **N-(hydroxyethy1)piperidines 14** and **15**  required for our purposes has been carried out by alkylation of the corresponding piperidines **21** and **22** with 2 bromoethanol following our usual procedure. $2a,8$  In turn, the starting piperidines **21** and **22** were prepared by the p-TsOH cyclization of the iminoacetals **19** and **20,** resulting from the condensation of **l-(phenylsulfonyl)indole-3-car**baldehyde (16)<sup>9</sup> and the appropriate primary amines  $17<sup>2b</sup>$ 





and 18,<sup>2c</sup> respectively. It is worth commenting that the cyclization of imine **20** furnished a 1:2 mixture of cis and trans isomers **22** and **23,** respectively, probably due to a minor difference of stability with respect to the 5-ethyl derivative isomers, whose trans form was the only one observed.<sup>2a</sup>

The KtBuO treatment of alcohol **14** furnished the expected 3-spiroindolenine 12,<sup>10</sup> the formation of which was monitored by **TLC** and demonstrated by **LiAlH,** reduction. Thus, a 1.2:l mixture of 3-spiroindolines **24** and **25** was obtained." They were separated by column chromatog-

(10) BSpiroindoleninea **are** usually prepared by the following methods (i) reduction of  $2$ -oxindoles,<sup>11</sup> which in turn are majorly obtained from chloroindolenines derived from indolo[2,3-a]quinolizidine systems by 'BuClO treatment;<sup>7b,12</sup> (ii) as a consequence of a retro-Mannich reaction indoles<sup>15</sup> or reaction with N-acyliminium ions.<sup>16</sup><br>(11) Jackson, A. H.; Smith, P. *Tetrahedron* 1968, 24, 2227-2239. "BuCIU treatment;""<sup>12</sup> (ii) as a consequence of a retro-Mannich reaction<br>in acid conditions on *8*-carbolines:<sup>13,14</sup> (iv) by alkylation of 3-substituted

**<sup>(8)</sup>** Alkylation of cis-piperidine 9 with 2-bromoethanol in several ex perimental conditions was unfruitful.

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### Applications of Protected 2-Aryl-4-piperidones

raphy and identified and differentiated by their spectroscopic data (see Tables III and IV). The most valuable information for identifying the spiroindolines was given by the presence of signals at ca. *6* 54, due to the spiro carbon, a doublet at 6 **71.5** corresponding to C-8, and the methylene carbon C-9 signal at  $\delta$  57-59 in the <sup>13</sup>C NMR spectra. In addition, the most characteristic data for differentiating the isomers was the chemical shift of 14-H in the <sup>1</sup>H NMR spectra, which appears at  $\delta$  7.28 when next to the piperidine nitrogen atom lone pair (in **25)** and at **6** 7.10 when next to 8-H and far away from the influence of the nitrogen lone pair (in **24).'\*** Another distinctive signal corresponds to the AB system observable for C-9 methylene protons, which appears **as** two doublets at 6 3.24 and 3.80 in **24** and as a singlet at 6 3.44 in **25.** The complete signal assignment and the conformational study was accomplished on the basis of 2D NMR data (TOCSY and NOESY) and is shown in Tables I11 and IV.

The rearrangement of indolenine  $12$  in  $BF_3$  $Et_2O$  led satisfactorily to the expected indolo[2,3-a]quinolizidine **6,19**  which was identified by comparison of its spectral data to those previously obtained.<sup>2b</sup>

The reaction course in aqueous **50%** AcOH implies the formation of an intermediate iminium salt **32,** which is converted into lactam **3l2I** or into enaminone **30,** when the acetal function is hydrolized. The formation of **32** in the aqueous acid medium was confirmed by its trapping with KCN to the corresponding  $\alpha$ -aminonitrile 33.<sup>22</sup> The conversion of **33** into the **indolo[2,3-a]quinolizidin-2-one**  ethylene acetal **6** was accomplished by means of dry *p-*TsOH treatment.

Finally, the direct conversion of  $\alpha$ -aminonitrile 33 into **indolo[2,3-a]quinolizidin-2-one (2)** was carried out by aqueous 50% AcOH treatment.<sup>23,24</sup>

In view of these satisfactory results, the application of this reaction sequence to preparation of l-ethylindolo- **[2,3-a]quinolizidin-2-one (1)** from alcohol **15** was envisaged. The KtBuO cyclization of alcohol **15** provided the corresponding 3-spiroindolenines 13, which on LiAlH<sub>4</sub> reduction were converted into a 2.5:2:1 mixture of 3-spiroindolines **26-28** (Scheme 111). Their structure and stereochemistry were once again determined from the spectral data.<sup>25</sup>

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(24) Indolo[2,3-a]quinolizidin-2-one 2 was also obtained from en-



Thus, in the <sup>13</sup>C NMR spectra, the signals at ca.  $\delta$  54 characteristic of the spiro-carbon C-14b, and at  $\delta$  42-45 and 72-76, corresponding to the ethyl-substituted C-7 and the angular C-8 carbons, respectively, together with the methylene carbon C-9 at  $\delta$  56-61, were important for determination of the structures (see Table IV). In the 'H NMR spectra, the chemical shift of the signal corresponding to 14-H allowed the stereochemical assignment on the spiro carbon **(see** Table In). **Thus,** in spiroindolines 27 and 28 this proton appeared at  $\delta \sim 7.1$ , while the third isomer **26** showed a deshielding to 6 7.33 due **to** the spatial proximity of this proton and the electron lone pair of the nitrogen atom. Furthermore, the disposition of the ethyl substituent was inferred from the 13C **NMR** data, **as** isomer **28** showed an upfield shift for C-5 and C-l4b, corresponding to a  $\gamma$ -gauche effect exerted by the ethyl group when axially disposed.

Treatment of C-7 epimeric acetals **26** and **27** with aqueous 10% HC1 afforded in both cases the piperidone **29** in good yields, **as** a consequence of a C-7 epimerization in the acidic conditions. $27$ 

When 3-spiroindolenines 13 reacted with  $BF_3E_2O$ , only enaminone **35** was obtained, resulting from the aqueous workup of the intermediate iminium salt **34,** formed because the steric hindrance due to the ethyl group prevents the rearrangement to the corresponding indolo[2,3-a] quinolizidine **8** and favors accumulation of the iminium salt. Enaminone **35** was identified by its very characteristic <sup>1</sup>H NMR spectrum, which presents a singlet at  $\delta$  6.65 for the olefinic proton, two signals at  $\delta$  0.83 and 2.00 corresponding to the ethyl substituent, and four triplets at  $\delta$ 2.40,3.02,3.38, and 3.49 due to the methylene protons of the molecule assigned to 5-H, In-CH<sub>2</sub>, 6-H, and NCH<sub>2</sub>, respectively. Signals at  $\delta$  228.3, 153.0, and 110.9 in the <sup>13</sup>C NMR spectrum confirmed the presence of a conjugated carbonyl function.

Finally, aqueous acid treatment of enaminone **35** led to the expected **l-ethylindolo[2,3-a]quinolizidin-2-one (1)** in **55%** yield. The direct transformation of spiroindolenines **13** into our target molecule **1** was **as** well accomplished, in 60% yield, by heating in an aqueous  $20\% \text{ H}_2\text{SO}_4$  medium.

An alternative and rather surprising way to reach our goal was found to be the aqueous acid (4 N HC1 or 20% H2S04) treatment of **7-ethyl-6,6-(ethylenedioxy)-**  1,2,5,6,7,7a-hexahydropyrido[1',2':1,2]pyrazino[4,3-*a*]indole<br>11, obtained by us in previous work,<sup>2b</sup> which implied a new kind **of** rearrangement, not observed in the case of unsubstituted systems. The formation of l-ethylindolo[2,3 a]quinolizidin-2-one **(1)** can be accounted for by considering that in the acid medium the indole 3-position is

**<sup>(18)</sup> The same 'H NMR effect is characteristic in the case of 2-oxindoles: (a) Crabb, T. A. In** *Annual Reports on NMR Spectroscopy;* **Webb,**  *G.* **A., Ed.; Academic Press: London, 1978. (b) Gaskell, A. J.; Randuz, H.-E.; Winterfeldt, E.** *Tetrahedron* **1970,26, 5353-5360.** 

**<sup>(19) 3-</sup>Spiroindolenines has been observed as intermediates to 2,3 disubstituted indoles through a Wagner-Meerwein** type **rearrangement,'6**  and the migratory aptitudes of diverse substituents has as well been **established.** 

**<sup>(20)</sup> Jackson, A. H.; Naidoo, B.** *Tetrahedron* **1968, 24, 6119-6129. (21) For a related formation of lactams as byproducts in** imini,um **salts**  cyclizations, see: (a) Rubiralta, M.; Torrens, A.; Palet, A.; Grierson, D.<br>S.; Husson, H.-P. *Tetrahedron Lett.* 1989, 30, 6761–6764. (b) Ponglux,<br>D.; Wongseripipatana, S.; Aimi, N.; Nishimura, M.; Ishikawa, M.; Sada,<br> **For a related formation of lactams in the mercuric acetate oxidation of piperidines, see: Fujii, T.; Ohba, M.; Sasaki, N.** *Heterocycles* **1984,22, 1805-1810.** 

**<sup>(22)</sup> Grierson, D. S.; Vuilhorgne, M.; Husson, H.-P.; Lemoine, G.** *J. Org. Chem.,* **1982,47,4439-4452. (23) (a) Lounasmaa, M.; Jokela, R.** *Tetrahedron* **1989,45,7449-7458.** 

**<sup>(24)</sup> Indolo[2,3-a]quinolizidin-2-one 2 was also obtained from en- aminone 30 (Winterfeldt, E.** *Chem. Ber.* **1964,2463-2468) as well as from lactam 31: Fujii, T.; Yoshifuji, S.; Ito, H.** *Chem. Pharm. Bull.* **1988, 36, 3348-3353.** 

<sup>(25)</sup> The stereochemical assignment of the major spiroindoline 26, in which the C-9 and C-14b bond is cis with respect to the nitrogen lone pair, corresponding to the "A series"<sup>26</sup> and the ethyl substituent is equatorial, series  $(\Delta \delta 2.5)$  and C-5 c.a. 3 ppm shilded when the ethyl chain is axial **due to a 'y-gauche" effect.** 

**<sup>(26)</sup> Finch, N.; Taylor, W. I.** *J. Am. Chem. SOC.* **1962,34,3871-3877.** 

**<sup>(27)</sup> For a related spiroindolizidm-7-one,** *see:* **Ban, Y.; &to, M.; Obhi, T.** *Chem. Pharm. Bull.* **1976,23,2605-2613.** 

protonated to give 36. The hindrance due to the proximity of the ethyl substituent upon the  $N_b$  electron lone pair makes its protonation difficult and therefore the  $N_b$  nitrogen atom can act as a nucleophile upon **C-7** with opening of ring C. The resulting aziridinium salt **37** is then opened by the indole from its most reactive 3-position (see Scheme V).

In conclusion, we can state that l-ethylindolo[2,3-a] quinolizidin-Pone **(1)** can be successfully prepared by **an**  intramolecular cyclization of N-(2-hydroxyethyl)-2-[1-**(phenylsulfonyl)-3-indolyl]-4-piperidone 4 as** well **as** from the 2-indolyl analogue<sup>2b</sup> by the action of K<sup>t</sup>BuO, followed by acid treatment. However, the new strategy described in the present paper is a one-pot reaction that provides a superior global yield (60%) with respect to the previous one<sup>2b</sup> (38%), which proceeds in two steps. Furthermore, the reactivity of 3-spiroindolenines will be developed toward other potential biologically active families of indole alkaloids.

### **Experimental Section**

General Methods. Melting points were determined in a ca- pillary tube on a CTP-MP 300 hot-plate apparatus and are uncorrected. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Varian XL-200 or a Varian Gemini-200 instrument. Unless otherwise noted, NMR spectra were registered in CDCl,, and chemical **shifts** are expressed in parts per milion **(6)** relative to internal Me4Si. IR spectra were recorded on a Perkin-Elmer 1430 spectrophotometer. Mass spectra were determined on a Hewlett-Packard 5988A mass spectrometer. Flash column chromatography was carried out on  $SiO<sub>2</sub>$  (silica gel 60, 40–63  $\mu$ m, Macherey–Nagel) or Al<sub>2</sub>O<sub>3</sub> (aluminum oxide 90 neutral, activity 1, 63-200  $\mu$ M, Merck). TLC was performed on  $SiO<sub>2</sub>$  (silica gel 60 F254, Merck) and developed with the solvent described in each case for flash chromatography. The spots were located by UV light and ioeffected according to standard methods. Prior to concentration under reduced pressure, all extracts were dried over anhydrous  $Na<sub>2</sub>SO<sub>4</sub>$  powder. Microanalyses were performed on a Carlo Erba 1106 analyzer by the Departament de Quimica Orgànica i Biolbgica, CSIC, Barcelona.

24 **l-(Phenylsulfonyl)-3-indolyl]-4-piperidone** Ethylene Acetal (21). A solution of amino acetal  $17<sup>2b</sup>$  (5.92 g, 45.2 mmol) and 1-(phenylsulfonyl)indolyl-3-carbaldehyde<sup>9</sup> (14.17 g, 49.7 mmol) in anhydrous benzene (200 mL) was stirred at 0  $^{\circ}$ C for 30 min, at room temperature overnight, and at reflux for 3 h. After 16 h of additional reflux with water removal by a Dean-Stark trap, the solvent was evaporated to give imine 19 (17.5 g, 98%), which was used without further purification.

A stirred mixture of the imino acetal 19 (17.5 g, 44.5 mmol) and anhydrous p-TsOH (15.23 g, 88.6 mmol) in dry benzene (400 mL) was heated at reflux under  $N_2$  for 1 h. The cooled mixture was washed with aqueous  $Na<sub>2</sub>CO<sub>3</sub>$ , dried, and evaporated to give a thick oil that was flash chromatographed  $(CH_2Cl_2-MeOH (95:5))$ to yield piperidine 21 (12.63 g, 72%) as a yellow solid: mp 93-94 OC (acetone); IR (KBr) 3310, 1370, 1170 cm-'; MS *m/e* (relative intensity) 398 (M<sup>+</sup>, 15), 353 (21), 283 (20), 257 (24), 227 (11), 197 (161, 171 (30), 142 (36), 130 (19), 115 (37), 87 (64), 77 (loo), 51 (22). Anal. Calcd for  $C_{21}H_{22}N_2O_4S \cdot 1/2H_2O$ : C, 61.90; H, 5.69; N, 6.87. Found: C, 61.78; H, 5.49; N, 6.85.

3-Ethyl-2-[ **l-(phenylsulfonyl)-3-indolyl]-4-piperidone**  Ethylene Acetals (22 and 23). Operating **as** above, from 1- **(phenylsulfonyl)indolyl-3-carbaldehydeg** (14.51 g, 50.9 mmol) and amino acetal  $18^{2b}$  (7.36 g, 46.29 mmol) in anhydrous benzene (200 **mL)** was obtained imine 20 (22.6 g, 98%), which, without further purification, was treated with anhydrous p-TsOH (15.6 g, 90.7 mmol), thus leading to a 1:2 mixture of trans- and cis-piperidines 22 and 23, which were separated by flash chromatography  $(CH_2Cl_2-MeOH (95:5))$ . cis-Isomer 23 (higher  $R_f$ , 5.2 g, 24%): mp 79–82 °C (acetone); IR (NaCl) 3300, 1360, 1170 cm<sup>-1</sup>; MS  $m/e$ (relative intensity) 426 (M<sup>+</sup>, 1), 382 (2), 396 (1), 312 (1), 283 (4), 270 (1), 115 (15), 77 (100). Anal. Calcd for C<sub>23</sub>H<sub>23</sub>N<sub>2</sub>O<sub>4</sub>S: C, 64.77; H, 6.14; N, 6.57. Found: C, *64.50;* H, **5.99;** N, 6.48. trans-Isomer

22 (lower *R,,* 9.93 g, 46%); mp *80-83* 'C (acetone); **IR** (NaCl) 1340, 1150 cm-'; CIMS *m/e* (relative intensity) 427 **(M+** + 1, loo), 304 (10), 287 (69), 213 (4), 177 (50), 125 (11). Anal. Calcd for C<sub>23</sub>H<sub>28</sub>N<sub>2</sub>O<sub>4</sub>S: C, 64.77; H, 6.14; N, 6.57. Found: C, 64.52; H, 6.33; N, 6.09.

N-(2-€Iydroxyethyl)-2-[ **l-(phenylsulfonyl)-3-indolyl]-4**  piperidone Ethylene Acetal (14). 2-Bromoethanol (1.34 mL, 18.84 mmol) was added dropwise to a mixture of piperidine 21  $(5 g, 12.56 mmol)$  and anhydrous  $K_2CO_3$  (5 g) in absolute ethanol (150 mL). The resulting mixture was heated at reflux under  $N_2$ for 15 h. The EtOH was evaporated, and the reaidue was dissolved in  $CH_2Cl_2$  and washed with  $H_2O$ . The dried organic phase was evaporated and purified by flash chromatography  $(Al_2O_3, CH_2Cl_2-MeOH (97:3))$  to give pure alcohol 14 (3.44 g, 62%) as a yellow solid: 96-97 °C (acetone); IR (KBr) 3550-3300, 1370 and 1175 cm<sup>-1</sup>; MS  $m/e$  (relative intensity) 442 (M<sup>+</sup>, 0.1), 411 (15), 353 (3), 296 (4), 283 (3), 257 (4), 215 (4), 142 (21), 128 (91), 115  $(24)$ , 99  $(64)$ , 77  $(100)$ , 42  $(86)$ . Anal. Calcd for C<sub>23</sub>H<sub>28</sub>N<sub>2</sub>O<sub>5</sub>S: C, 62.43; H, 5.92; N, 6.33. Found: C, 62.49; H, 5.79; N, 6.46.

trans -3-Ethyl- 1-(2-hydroxyethyl)-2-[ 1-( phenyl**sulfonyl)-3-indolyl]-4-piperidone** Ethylene Acetal (15). Operating **as** above, from tram-piperidine 22 (2.2 g, 5.16 mmol), anhydrous  $K_2CO_3$  (2 g), absolute EtOH (200 mL), and 2bromoethanol (0.55 mL, 7.74 mmol) was obtained alcohol 15 after flash chromatography  $(Al_2O_3, CH_2Cl_2-MeOH (97:3))$  purification, together with the corresponding trams -1-(2-hydroxyethyl)-2- **(3-indolyl)-4-piperidone** ethylene acetal (lower *Rf,* 210 mg, 12%): mp 119-121 °C (acetone); IR (KBr) 3350-3200  $cm^{-1}$ ; CIMS *m/e* (relative intensity) 331 (M+ + 1, loo), 287 (14), 207 (12), 180 (18), 163 (21). Anal. Calcd for  $C_{19}H_{26}N_2O_3.1/2H_2O$ : C, 67.84; H, 8.00; N, 8.33. Found: C, 67.89; H, 7.81; N, 8.12. Alcohol 15 (higher *R<sub>f</sub>*, 1.69 g, 70%): IR (NaCl) 3300-3200, 1360, 1170 cm<sup>-1</sup>; MS  $m/e$  (relative intensity) 470 (M<sup>+</sup>, 1), 439 (10), 329 (1), 298 (1), 128 (63), 99 (46), 77 (100), 42 (33). Anal. Calcd for  $C_{25}H_{30}N_2O_5S$ : C, 63.80; H, 6.43; N, 5.95. Found: C, 63.53; H, 6.89;

N, 5.64.<br>General Method To Prepare 3-Spiroindolenines. To a solution of the amino alcohols 14-15 (1 equiv) in anhydrous THF was added freshly sublimed K<sup>t</sup>BuO (2 equiv) under a N<sub>2</sub> atmosphere, at  $0 °C$ . The mixture was stirred for  $30$  min at  $0 °C$ , and a TLC control of the indolenine formation was performed  $(SiO<sub>2</sub>,$  $CH<sub>2</sub>Cl<sub>2</sub>–MeOH (95:5)$ ).

**7-Oxoindolizidme-l-spiro-3'-indoline** Ethylene **Acetals (24**  and 25). To a solution of spiroindolenine 12, prepared **as** in the general methcd from alcohol 14 (300 **mg,** 0.68 mmol), in *dry* THF (30 **mL)** was added LiAlH4 (51.5 **mg,** 1.46 mmol), and the mixture was heated at reflux for 15 min under a  $N_2$  atmosphere. The reaction was quenched with saturated aqueous NH4Cl (1 **mL),**  and the reaction mixtures was poured into ice-water and  $Et<sub>2</sub>O$ . The layers were separated, and the aqueous phase was extracted twice with  $Et_2O$  and one time with  $CH_2Cl_2$ . The mixed organic extracts were dried and evaporated to give a 1:1 epimeric mixture of 3-spiroindolenines 24 and 25, which were separated by flash chromatography (CH<sub>2</sub>Cl<sub>2</sub>-MeOH (93:7)). Spiroindoline 24 (higher *Rf,* 48.6 mg, 25%): IR (NaC1) 3300-3400 cm-'; MS *m/e* (relative intensity) 286 (M<sup>+</sup>, 3), 156 (100), 144 (12), 130 (17). Spiroindoline 25 (lower  $R_f$ , 58.35 mg, 30%): IR (CHCl<sub>3</sub>) 3400 cm<sup>-1</sup> cm<sup>-1</sup>; MS *m/e* (relative intensity (286 (M+, 4), 156 (loo), 144 (9), 130 (12). Anal. Calcd for  $C_{17}H_{22}N_2O_2$ : C, 71.30; H, 7.74; N, 9.78. Found: C, 70.98; H, 7.87; N, 9.32.

**8-Ethyl-7-oxoindolizidine-l-spiro-3'-indoline** Ethylene Acetals (26-28). Operating **as** above, from a solution of spire indolenine 13, prepared following the general procedure from 15 (300 mg, 0.64 mmol) in dry THF (30 mL), and **LiAlH4** (49 mg, was separated by flash chromatography (CH<sub>2</sub>Cl<sub>2</sub>-MeOH (95:5)). Spiroindoline 26 *(R,* 0.54,42 *mg,* 21%): IR (NaCl) 3350 cm-'; **MS**  *m/e* (relative intensity) 314 (M<sup>+</sup>, 1), 227 (2), 184 (100), 168 (22), 130 **(23),** 117 (17), 98 (14). Spiroindoline 27 *(Rf0.25,* 32 *mg,* 16%). Spiroindoline 28 (0.38, 16 mg, 8%): Anal. Calcd for H, 8.50; N, 8.83.  $C_{19}H_{26}N_2O_2·1/4H_2O$ : C, 71.49; H, 8.31; N, 8.78. Found: C, 71.72;

**8-Ethyl-7-oxoindolizidine-** l-spiro-3'-indoline (29). A *so*lution of spiroindoline ethylene acetal 26 of 27 (50 mg, 0.16 mmol) in a 1:l mixture of 4 N HC1 and MeOH (40 mL) waa heated at reflux for 24 h. The reaction mixture was cooled, basified with

### Applications of Protected 2-Aryl-4-piperidones

NazC03 and extracted with CHzC1z to provide ketone **29** (36 mg, 89% from 26) after flash chromatography (CH<sub>2</sub>Cl<sub>2</sub>-MeOH (95:5)): IR (NaC1) 1695 (CO) cm-'; MS *m/e* (relative intensity) 270 (M', 34), 257 (4), 140 (100), 117 (14), 110 (12), 82 (19). Anal. Calcd for  $C_{17}H_{22}N_2O$ : C, 75.51; H, 8.20; N, 10.36. Found: C, 75.13; H, 8.08; N, 10.41.

trans **-2,2- (Et hy1enedioxy)- 1,2,3,4,6,7,12,1 Bb-octahydro- [2,3-a]quinolizine (6). Method A.** To a solution of spiroindolenine **12,** prepared **as** in the general procedure from alcohol **14** (300 mg, 0.68 mmol), in dry THF (30 mL) was added freshly distilled  $BF_3-Et_2O$  (0.126 mL, 1.02 mmol), and the resulting mixture was treated at reflux for 2 h under  $N_2$  atmosphere. The reaction mixture was poured into ice-water, basified with  $Na<sub>2</sub>CO<sub>3</sub>$ , and extracted twice with  $Et_2O$  and one time with  $CH_2Cl_2$ . The organic extracts were dried and evaporated to yield an oil that was purified by flash chromatography ( $CH_2Cl_2-MeOH$  (95:5)), thus isolating pure indolo<sup>[2,3-a]quinolizidin-2-one ethylene acetal</sup> **6;** 75 mg, 40%, which was identified by comparison of its spectral data to the described ones.<sup>2b</sup>

**Method B.** A solution of  $\alpha$ -aminonitrile 33 (400 mg, 1.29 mmol) in dry benzene (25 mL) was added to a solution of anhydrous p-TsOH (47 mg, 2.6 mmol) in dry benzene (75 mL), and the mixture was treated at reflux for 1 h under a  $N_2$  atmosphere. The reaction was poured into ice-water and basified with  $Na<sub>2</sub>CO<sub>3</sub>$ . The layers were separated, and the aqueous phase was extracted with benzene. The mixed organic layers were dried and evaporated to give an oil that was flash chromatographed  $\rm (CH_2Cl_2-MeOH)$ (955)) to yield pure **indolo[2,3-a]quinolizidine 6** (147 mg, 40% ), which was identified by comparison of its spectral data with those previously obtained.<sup>26</sup>

**Treatment** of **Spiroindolenine 12 with 50% Aqueous**  was added 50% aqueous AcOH (30 mL), and the resulting mixture was stirred at room temperature for 30 min. The reaction mixtures was basified with  $Na<sub>2</sub>CO<sub>3</sub>$ , the layers were separated, and the aqueous phase was extracted twice with  $Et<sub>2</sub>O$  and one time with  $CH<sub>2</sub>Cl<sub>2</sub>$ . The organic extracts were dried and evaporated to give an oil that was flash chromatographed  $(CH_2Cl_2-MeOH (95:5)),$ thus isolating **4,4-(ethylenedioxy)-l-tryptophyl-2-piperidone (31)** (higher  $R_f$ , 17 mg, 9%): IR (CHCl<sub>3</sub>) 3440, 1650 cm<sup>-1</sup>; <sup>1</sup>H RMN: 1.82 (m, 2 H, 5-H), 1.90-2.70 (m, 6 H, 3-H, 5-H and In-CH<sub>2</sub>), 3.24 (dt,  $J = 6$  and 2 Hz, 2 H, NCH<sub>2</sub>), 3.56 (td,  $J = 10$ and 4 Hz, 2 H, 6-Ha), 3.90 (br d,  $J = 10$  Hz, 1 H, 6-He), 3.96-3.98 (m, 4 H, OCH<sub>2</sub>), 7.10-7.50 (m, 2 H, In-5H and In-6H), 7.42 (s, 1 H, In-2H), 7.60-8.00 (m, 2 H, In-7H and In-4H), 9.20-9.30 (br **8,** 1 H, In-NH); 13C NMR 24.1 (In-CH2), 41.5 (C-5), 49.8 (C-31, 54.5 (C-6), 58.0 (NCH<sub>2</sub>), 64.3 (OCH<sub>2</sub>), 100.6 (C-4), 111.9 (In-C7), 114.0 (In-C3), 120.3 (In-C5), 121.8 (In-CG), 123.0 (In-C2), 129.3 (In-C3a), 136.5 (In-C7a), 185.7 (C<del>=</del>O); MS  $m/e$  (relative intensity) 300 (M', 25), 284 (loo), 283 (75), 253 (l), 239 (E), 197 (57), 145 (57). Anal. Calcd for  $C_{17}H_{20}N_2O_3$ : C, 67.98; H, 6.71; N, 9.33. Found: C, 67.59; H, 6.83; N, 9.01.

**l-Triptophyl-2,3-dihydro-lH-pyridin-4-one (30)** (lower *R,*  80 mg, 49%): IR (CHCl<sub>3</sub>) 3480, 1640, 1600 cm<sup>-1</sup>; <sup>1</sup>H NMR 2.38  $(t, J = 8$  Hz, 2 H, 3-H), 3.01  $(t, J = 7$  Hz, 2 H, In-CH<sub>2</sub>), 3.46  $(t, J = 8)$  $J = 8$  Hz, 2 H, 2-H), 3.51 (t,  $J = 7$  Hz, 2 H, In-CH<sub>2</sub>CH<sub>2</sub>), 4.77 (d,  $J = 7$  Hz, 1 H, 5-H), 6.76 *(d, J = 7 Hz, 6-H)*, 6.97 *(s, 1 H, In-2H)*, 7.10 (t, J = 7 Hz, 1 H, In-5H), 7.19 (t, J = 7 Hz, 1 H, In-6H), 7.38 (br *s*, 1 H, In-NH); <sup>13</sup>C NMR δ 24.6 (In-CH<sub>2</sub>), 34.5 (C-3), 46.4 (C-2), 56.2 (In-CH<sub>2</sub>CH<sub>2</sub>N), 96.2 (C-5), 110.5 (C-4), 111.6 (In-C7), 117.9 (In-C2), 119.1 (In-C5), 121.8 (In-C4), 122.9 (In-C6), 123.1 (In-C3), 126.6 (In-C3a), 136.4 (In-C7a), 155.5 (C-6), 192.1 (C=O); MS  $m/e$ (relative intensity) 240 (M', 25), 143 (6), 130 (loo), 111 **(55),** 110 (41), 82 (23). The hydrochloride melted at  $160-162$  °C (acetone-MeOH) (lit.<sup>21</sup> mp 164 °C).  $(d, J = 7 Hz, 1 H, In-7H)$ , 7.53  $(d, J = 7 Hz, 1 H, In-4H)$ , 9.45-9.55

**Treatment of Spiroindolenine 12 with Aqueous KCN. To**  a solution of spiroindolenine **12** prepared as above was added a solution of KCN (133 mg, 2.05 mmol) in an aqueous buffer solution (pH 4) of citric acid-sodium acetate (30 mL), and the mixture was vigorously stirred at room temperature for 30 min. The emulsion was basified with  $Na<sub>2</sub>CO<sub>3</sub>$ , the layers were separated, and the aqueous phase was extracted twice with  $Et_2O$  and one time with  $CH_2Cl_2$ . The mixed organic extracts were dried and evaporated to give an oil that was flash chromatographed  $(CH<sub>2</sub>Cl<sub>2</sub>-MeOH (95:5))$ , thus obtaining  $\alpha$ -aminonitrile 33 (110) mg, 69%): IR (CHCl<sub>3</sub>) 3480, 2210 cm<sup>-1</sup>; <sup>1</sup>H NMR δ 1.70-2.00 (m, 1 H, 5-H), 2.60-3.00 (m, 1 H, 5-H), 3.85-4.10 (m, **5** H, OCHz and 2-H), 7.01 (d,  $J = 1.5$  Hz, 1 H, In-2H), 7.13 (br t,  $J = 7$  Hz, 1 H, In-5H), 7.14 (br d,  $J = 7$  Hz, 1 H, In-6 H), 7.32 (br d,  $J = 7$  Hz, **<sup>1</sup>**H, In-4H), 7.62 (br d, J = 7 Hz, 1 H, In-7H), 8.30 (br s, 1 H, NH); <sup>13</sup>C NMR 22.9 (In-CH<sub>2</sub>), 34.4 (C-5), 36.5 (C-3), 47.2 (C-6), 50.8 (C-2), 55.7 (In-CH<sub>2</sub>CH<sub>2</sub>), 64.2 and 64.5 (OCH<sub>2</sub>), 105.3 (C-4), 111.2 (In-C7), 113.2 (In-C3), 117.0 (CN), 118.5 (In-C4), 119.2 (In-C5), 121.8 and 121.9 (In-C6 and In-C2), 127.3 (In-CBa), 136.2 (In-C7a); MS  $m/e$  (relative intensity) 311 (M<sup>+</sup>, 12), 181 (100), 154 (62), 130 (38), 82 (10). Anal. Calcd for  $C_{18}H_{21}N_3O_2$ : C, 69.43; H, 6.83; N, 13.50. Found: C, 69.50; H, 7.04; N, 13.71.

*trans* **-3,4,6,7,12,12bHexahydro[ 2,343 Iquinolizin-2( lH)-one (2).** A solution of a-aminonitrile **33** (100 mg, 0.32 mmol) in MeOH temperature for 15 h. The reaction mixtures was basified with Na<sub>2</sub>CO<sub>3</sub> and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic extracts were dried and evaporated, and the oil thus obtained was flash chromatographed (CHZClz-MeOH (955)) yielding **2 (50** mg, 65%), which was identified by comparison of its spectral data to the previously obtained.<sup>2b</sup>

trans **-1 -Ethyl-3,4,6,7,12,12b-hexahydroindolo[2,3-a 1 quinolizin-2(lH)-one** (1). **Method A.** Operating as for the preparation of **6,** from a solution of spiroindolenine **13,** prepared as in the general method from alcohol **15** (300 mg, 0.64 mmol) in dry THF (30 mL), and freshly distilled  $BF_3E_6O$  (118 mL, 0.96 mmol) was obtained enaminone **35** (92 mg, 46%) after flash chromatography (CH<sub>2</sub>Cl<sub>2</sub>-MeOH (95:5)): IR (CHCl<sub>3</sub>) 3460, 1590 cm<sup>-1</sup>; <sup>1</sup>H NMR 0.83 (t,  $J = 7$  Hz, 3 H, CH<sub>3</sub>CH<sub>2</sub>), 2.00 (q,  $J = 7$ 2 H, In-CH<sub>2</sub>), 3.38 (t,  $J = 8$  Hz, 2 H, 6-H), 3.49 (t,  $J = 7$  Hz, 2 H, In-CH<sub>2</sub>CH<sub>2</sub>), 6.65 (s, 1 H, 2-H), 6.95 (s, 1 H, In-2H), 7.10 (t,  $J = 7$  Hz, 1 H, In-4H), 7.18 (t,  $J = Hz$ , 1 H, In-5H), 7.35 (d, J  $= 7$  Hz, 1 H, In-7H), 7.55 (d,  $J = 7$  Hz, 1 H, In-4H), 8.70-8.90 (br s, 1 H, NH); <sup>13</sup>C NMR 14.3 ( $CH_3CH_2$ ), 20.1 ( $CH_3CH_2$ ), 24.1 (In-C7), 118.4 (In-C2), 119.5 (In-C5), 122.3 (In-C4), 123.1 (In-CG), 127.1 (In-Cia), 128.6 (In-C3), 136.7 (In-C7a), 153.0 (C-2), 228.3 (C=O); MS *m/e* (relative intensity) 268 (M', 24), 184 (l), 139 (44), 138 (100), 130 (68), 110 (17), 55 (7). Anal. Calcd for N, 10.79. Hz, 2 H, CH<sub>2</sub>CH<sub>3</sub>), 2.40 (t,  $J = 8$  Hz, 2 H, 5-H), 3.02 (t,  $J = 7$  Hz, **(In-CH<sub>2</sub>)**, 35.8 **(C-5)**, 47.2 **(InCH<sub>2</sub>CH<sub>2</sub>)**, 56.4 **(C-6)**, 110.9 **(C-3)**, 111.8  $C_{17}H_{20}N_2O$ : C, 76.09; H, 7.51; N, 10.44. Found: C, 76.35; H, 7.52;

A solution of enaminone **35** (40 mg, 0.15 mmol) in MeOH (10 mL) and aqueous 10%  $H_2SO_4$  (10 mL) was stirred at 90-95 °C for 6 h. The reaction mixture was poured into ice-water, basified with  $Na<sub>2</sub>CO<sub>3</sub>$ , and extracted with  $CH<sub>2</sub>Cl<sub>2</sub>$ . The organic extracts, dried and evaporated, furnished an oil that was purified by flash chromatography ( $CH_2Cl_2$ -MeOH (95:5)) to give quinolizidin-2-one 1 (22 mg,  $55\%$ ): IR (CHCl<sub>3</sub>) 3320, 2795, 2745, 1690 cm<sup>-1</sup>; <sup>1</sup>H NMR 0.77 (t,  $J = 7$  Hz, 3 H, CH<sub>2</sub>CH<sub>3</sub>), 1.20-1.40 (m, 1 H, CH<sub>A</sub>CH<sub>3</sub>), 1.60-1.80 (m, 1 H,  $CH_BCH_3$ ), 2.35 (br d,  $J = 12$  Hz, 1 H,  $3-He$ ), 2.56 (m, 1 H, 1-He), 2.54-2.58 (m, 1 H, 6-H $\alpha$ ), 2.68 (br t,  $J = 12$ Hz, 1 H, 3-Ha), 2.70-2.80 (m, 1 H, 4-Ha), 2.78 (br d,  $J = 12$  Hz, 1 H, 7-H $\alpha$ ), 2.98 (t, J = 12 Hz, 1 H, 7-H $\beta$ ), 3.30 (br t, J = 6 Hz, 1 H, 4-He), 3.68 (br s,  $W_{1/2} = 7$  Hz, 1 H, 12b-H), 7.14 and 7.17  $(2 t, J = 7 Hz, 1 H each, 9-H and 10-H)$ , 7.35 (d,  $J = 7 Hz, 1 H$ , 11-H), 7.49 (dd, J = 7 Hz, 1 H, 8-H), 7.90 (br *8,* 1 H, NH); 13C (C-6), 54.9 (C-4), 56.2 (C-l), 62.4 (C-l2b), 108.0 (C-7a), 111.2 (C-ll), 118.3 (C-9), 119.7 (C-8), 121.9 (C-lo), 126.6 (C-7b), 131.5 (C-lla), 136.0 (C-12a), 211.2 (C=O); MS  $m/e$  (relative intensity) 268 M<sup>+</sup>, 69), 267 (loo), 253 (87), 197 (21), 169 (48), 115 (21), **55** (32). Anal. Calcd for  $C_{17}H_{20}N_2O^{-1}/_2H_2O$ : C, 73.56; H, 7.63; N, 10.09. Found: C, 73.19; H, 7.36; N, 9.78. NMR 11.4 (CH<sub>2</sub>CH<sub>3</sub>), 20.3 (CH<sub>2</sub>CH<sub>3</sub>), 21.7 (C-7), 38.8 (C-3), 52.1

**Method** B. To a solution of spiroindolenine **13,** prepared **as**  in the general method from alcohol **15** (300 mg, 0.64 mmol) in anhydrous THF (30 mL), was added 20% aqueous  $H_2SO_4$  (30 mL), and the mixture was heated at reflux for 15 h. The reaction mixture was poured into ice-water, basified with  $Na<sub>2</sub>CO<sub>3</sub>$ , and extracted with  $CH_2Cl_2$ . The organic layer, dried and evaporated, provided an oil which, after flash chromatography  $(CH_2Cl_2-MeOH$ <br>(95:5)) gave indoloquinolizidinone 1 (103 mg, 60%), which was identified by comparison of its spectral data with those already obtained.

Method C. A solution of pyridopyrazinoindole 11<sup>2b</sup> (50 mg, 0.16 mmol) in a 1:l mixture of 4 N HCl (10 mL) and MeOH (10 mL) **was** heated at reflux for **4** h. The reaction mixtures was cooled, basified with Na<sub>2</sub>CO<sub>3</sub>, and extracted with CH<sub>2</sub>Cl<sub>2</sub> to provide quinolizidinone 7 (38 mg, 95%) after flash chromatography (CH<sub>2</sub>Cl<sub>2</sub>-MeOH (95:5)), which was identified by comparison of ita TLC and spectral data with those already obtained.

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Registry **No. 1, 132113-29-2; 2, 55854-97-2; 6, 130179-28-1; 15,132113-30-5; 15** (de-(phenylsulfonyl) derivative), **136061-57-9; 11,130179-30-5; 12,136061-54-6; 13,136061-557; 14,130627-359; 16,80360-20-9; 17,62240-37-3; 18,130179-18-9; 19,130627-33-7; 20,132113-31-6; 21,130627-34-8; 22,132113-33-8; 23,132113-32-7; 24,130627-38-2; 25,130627-37-1; 26,13220044-3; 27,132200454; 28,132113-34-9; 29,136061-56-8; 30,92579-46-9; 31,130627-39-3; 33, 130627-40-6; 35, 132113-35-0.** 

Supplementary Material Available: **2D NMR** spectra of pertinent compounds **(4** pages). Ordering information is given on any current masthead page.

# **Utilizing Acetyl Hypofluorite for Chlorination, Bromination, and Etherification of the Pyridine System**

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Acetyl hypofluorite, which is easily made from  $F_2$ , possesses a strong electrophilic fluorine. This electrophile is able to attach itself to the nitrogen atom of pyridine and activate the ring toward nucleophilic attack ultimate elimination of HF results in **an** overall easy nucleophilic displacement of the hydrogen of the important 2-position. The nucleophiles used;  $Cl^b$ , Br<sup>b</sup>, and  $RO^b$ , originate from solvents such as  $CH_2Cl_2$ ,  $CH_2Br_2$ , and various primary alcohols. Thus, 2-halo- or 2-alkoxypyridines were formed. The reaction conditions (room temperature, very short reaction times, and good yields) transform the task of direct substitution of the pyridine ring from an extremely difficult to a very easy procedure.

During the last few years we have demonstrated that, apart from its more obvious function as a fluorinating agent, F<sub>2</sub> can be used for an array of processes leading eventually to difficult to obtain, fluorine-free compounds. We have utilized reagents directly prepared from fluorine for introducing double bonds in deactivated saturated sites,<sup>1</sup> for bromination and iodination of aromatic compounds,<sup>2,3</sup> for hydroxylation of saturated tertiary C-H bonds,<sup>4</sup> and for efficient epoxidation of many types of olefins.<sup>5</sup>

The pyridine system is of course very important in organic and pharmaceutical chemistry. Despite numerous research reports dealing with this system, reactions aimed specifically at direct substitution of the parent hydrogen at the important 2-position are very rare. Hydroxylations through arrangements of the appropriate  $N$ -oxide<sup>6</sup> and Chichibabin's amination' are practically the only routes for activating this position. Direct regiospecific halogenation of the pyridine ring is extremely difficult and usually unsatisfactory, and yet, examination of the literature leads to the conclusion that halopyridines constitute a very large part of this heterocycle's chemistry? Recently we have discovered that acetyl hypofluorite (AcOF), made from  $F_2$ ,<sup>9</sup> can be used for direct acetoxylation of the pyridine ring by utilizing the strong electrophilicity of the oxygen-bound fluorine coupled with the formation of the very strong HF bond.<sup>10</sup> We present here a somewhat unexpected,<sup>11</sup> yet



general, reaction derived from the action of AcOF on the pyridine ring leading to chlorination, bromination, and alkoxylation of this relatively inactive heterocycle.12 The mild conditions of this reaction and its efficiency (a few seconds at room temperature and usually high yields) are unparalleled for this type of chemistry.13

We found that the outcome of applying AcOF to pyridine **(1)** depends on the solvent system used. **As** we have already reported,<sup>10</sup> use of apolar solvents such as CFCl<sub>3</sub> resulted in 2-acetoxypyridine **(2)** in excellent yield. With  $CH_2Cl_2$ , however, the reaction course is altered and the major product, formed in 70% yield, is 2-chloropyridine

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**<sup>(11)</sup> It 1s worth mentioning in this respect Umemoto's and Zupan's work, which, although different from ours, still beam some similarities:**  (a) Umemoto, T.; Tomizawa, G. Tetrahedron Lett. 1987, 28, 2705. (b) Stavber, S.; Zupan, M. Tetrahedron Lett. 1990, 31, 775.

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**<sup>(13)</sup> In many cases, substitution of the 2-hydrogen with some other group proceeds very slowly at very high temperatures, and in poor yields. Thus a reaction with KOH yields only traces of 2(1H)-pyridone, while**  autoclave treatment with CuSO<sub>4</sub> of some substituted derivatives, such as 3-picoline at 300 °C, gives the corresponding pyridones in less than 7% yield. Tomasik, P.; Woszczyk, A. *Tetrahedron Lett*. 1977, 2193.